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## Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$   
Disorder in main residue  
 $R$  factor = 0.047  
 $wR$  factor = 0.127  
Data-to-parameter ratio = 15.4For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.4-(2,3,4,6-Tetra-*O*-acetyl- $\beta$ -D-galacto-  
pyranosyl)thiosemicarbazide

In the title compound,  $\text{C}_{15}\text{H}_{23}\text{N}_3\text{O}_9\text{S}$ , the hexopyranosyl ring adopts a chair conformation. The acetyl group at C4 occupies an axial position, while all other substituents are equatorial. The molecules are linked by  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds into ribbons parallel to the  $a$  axis, and  $\text{N}-\text{H}\cdots\text{O}$  interactions provide further stability in a three-dimensional network.

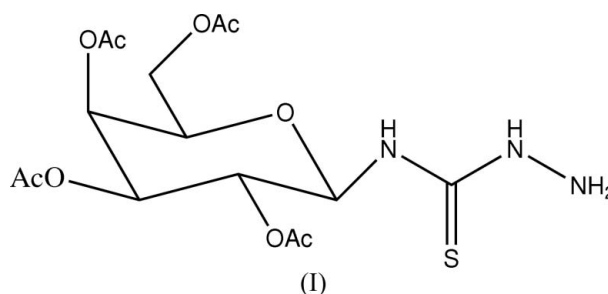
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## Comment

Substituted thioureas and thiosemicarbazides have attracted much attention in recent years because of their anti-HIV potential (Venkatachalam *et al.*, 2001) and their importance in the preparation of corresponding semicarbazides (Li *et al.*, 2001) and heterocyclic compounds (Wang *et al.*, 2001). We have reported the synthesis and structure of *N*-amino-*N*-(2,3,4,6-tetra-*O*-acetyl- $\beta$ -D-xylopyranosyl)thiocarbamide (II) (Yang *et al.*, 2004) and this work has been extended to the synthesis of the title compound (I) whose structure is reported here (Fig. 1, Table 1). All bond lengths and angles in (I) are within normal ranges (Allen *et al.*, 1987), and compare well with those values in compound (II). The acetyl group at C4 occupies an axial position, while all other substituents are equatorial. The pyranosyl ring adopts a chair conformation with the S1 atom in a synperiplanar position with respect to C1. The C1–N1–C15–S1 torsion angle is  $-5.5(4)^\circ$ . Atom O5 is disordered over two positions, with refined site occupancies of 0.32 (6) for O5A and 0.68 (6) for O5B.



In the crystal structure of (I), molecules are linked into ribbons along the  $a$  axis by  $\text{C}12-\text{H}12\text{C}\cdots\text{O}5\text{B}$  hydrogen bonds (Fig. 2). The packing is further stabilized by  $\text{N}2-\text{H}2\text{A}\cdots\text{O}1$  interactions (Table 2), forming a three-dimensional framework.

## Experimental

The title compound was prepared by the method of Yang *et al.* (2004). Colourless crystals suitable for X-ray crystallographic analysis were obtained by recrystallization from ethyl acetate/petroleum ether (1:2  $v/v$ ).

Crystal data

C<sub>15</sub>H<sub>23</sub>N<sub>3</sub>O<sub>9</sub>S  
*M<sub>r</sub>* = 421.42  
 Orthorhombic, *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>  
*a* = 7.5445 (11) Å  
*b* = 8.6792 (13) Å  
*c* = 31.229 (5) Å  
*V* = 2044.9 (5) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.369 Mg m<sup>-3</sup>

Mo *K*α radiation  
 Cell parameters from 3602 reflections  
 $\theta = 2.4\text{--}22.3^\circ$   
 $\mu = 0.21\text{ mm}^{-1}$   
*T* = 293 (2) K  
 Needle, colourless  
 0.30 × 0.14 × 0.11 mm

Data collection

Siemens SMART CCD area-detector diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.940, *T<sub>max</sub>* = 0.977  
 11659 measured reflections

4041 independent reflections  
 3407 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.024  
 $\theta_{\text{max}} = 26.0^\circ$   
*h* = -9 → 9  
*k* = -7 → 10  
*l* = -37 → 38

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.047  
*wR* (*F*<sup>2</sup>) = 0.127  
*S* = 1.02  
 4041 reflections  
 263 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0739P)^2 + 0.2979P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.22\text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.25\text{ e \AA}^{-3}$   
 Absolute structure: Flack (1983), 1690 Friedel pairs  
 Flack parameter: 0.03 (12)

Table 1

Selected geometric parameters (Å, °).

S1—C15	1.679 (3)	N1—C1	1.414 (3)
O1—C1	1.426 (3)	N2—C15	1.332 (4)
O1—C5	1.427 (3)	N2—N3	1.395 (4)
N1—C15	1.350 (3)		
N3—N2—C15—N1	5.7 (4)	C1—N1—C15—N2	175.9 (2)
N3—N2—C15—S1	-172.9 (2)	C1—N1—C15—S1	-5.5 (4)

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2A...O1 <sup>i</sup>	0.86	2.58	3.427 (3)	168
C12—H12C...O5B <sup>ii</sup>	0.96	2.31	3.12 (2)	141

Symmetry codes: (i)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $x + \frac{1}{2}, -y + \frac{1}{2}, -z$ .

All H atoms were positioned geometrically and treated as riding, with C—H = 0.96–0.98 Å [*U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C) or 1.5*U*<sub>eq</sub>(C) for the methyl H atoms] and N—H = 0.86 Å [*U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(N)].

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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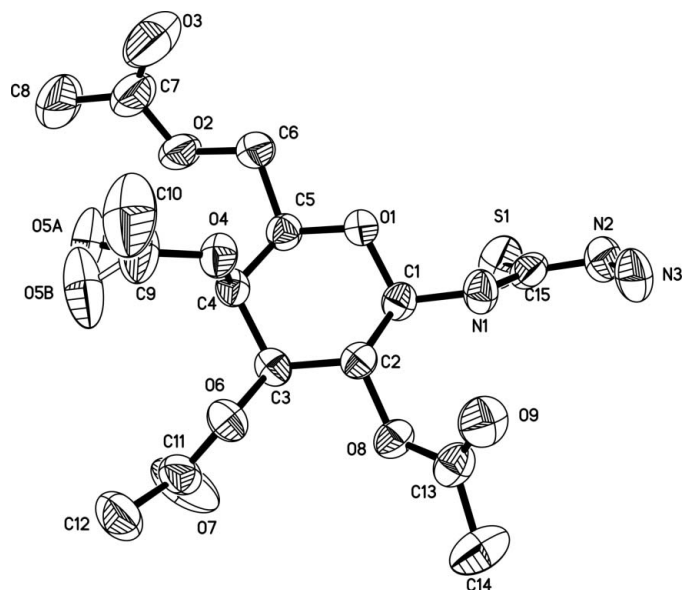


Figure 1

The structure of compound (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. H atoms are omitted for clarity.

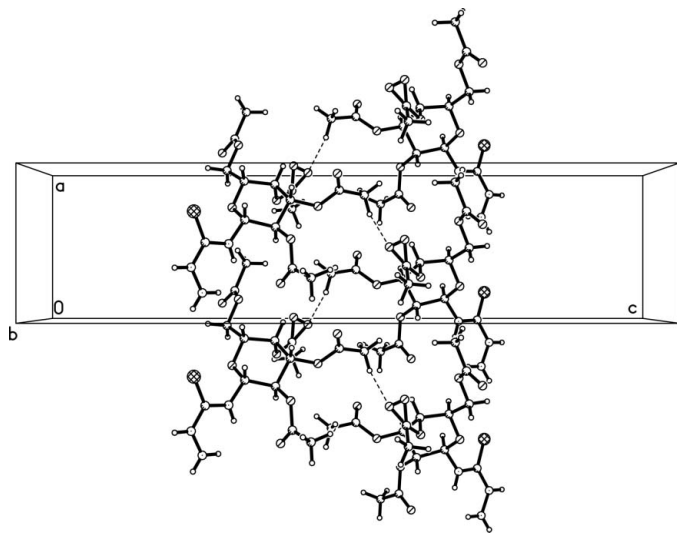


Figure 2

Packing diagram for (I), showing the formation of ribbons along the *a* axis. H bonds are drawn as dashed lines.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, S1–19.  
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
 Li, Z., Wang, X., Da, Y. & Chen, J. (2001). *Synth. Commun.* **31**, 1433–1440.  
 Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS, Inc., Madison, Wisconsin, USA.  
 Siemens (1996). *SMART* and *SAINTE*. Siemens Analytical X-Ray Systems, Inc., Madison, Wisconsin, USA.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
 Venkatachalam, T. K., Sudbeck, E. A., Mao, C. & Uckun, F. M. (2001). *Bioorg. Med. Chem. Lett.* **11**, 523–528.  
 Wang, X., Li, Z., Zhang, Z. & Da, Y. (2001). *Synth. Commun.* **31**, 1907–1911.  
 Yang, B., Zhang, S.-S., Wang, Y.-F., Li, X.-M. & Jiao, K. (2004). *Acta Cryst.* **E60**, o568–o570.